Cs₄PbBr₆-CsPbBr₃ Nanoheterostructures

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25M Emission intensity, counts PMAO Cs₄PbBr₆-CsPbBr₃ Olevlamine ligands Eu from the surface of nanocrystals show wavelength, I 400 440 rties ~35K Chemistry Cs₄PbBr₆ nanocrystals emission dual 513 nm at 320 -(CsPbBra with T~35 K under ~312 react а CO-polymer maleic nm excitation with an of map, Excitation v 20 360 (PMAO), anhydride indication of the 0 COO- NH 300 causing nanocrystal energy transfer (see CsPbBr Ω 376 nm PL (Cs₄PbBr₆) 320 Cs4PbBr6 transformation: paper for additional Cs₄PbBr₆ $Cs_{4}PbBr_{6} \rightarrow CsPbBr_{3}$ details). 400 440 480 520 360 Emission wavelength, nm High-Resolution Electron Microscopy (HRTEM) Transformation $Cs_4PbBr_6 \rightarrow CsPbBr_3$ under mild 1) conditions provides access to heterostructures, which **Significance** could be model systems for, e.g., green-emitting [or widegap] defects in Cs₄PbBr₆ [in CsPbBr₃]. Structure $C_{16}H_{33}$ 2) At low T, both components of Cs₄PbBr₆-CsPbBr₃ nanocrystals emit, 0、 creating opportunities for HO spectroscopic studies of the energy transfer between them, and novel R CsPbBr₂ single particle devices.

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